# NOTES

# Stevastelin A3, D3 and E3, Novel Congeners from a High Producing Mutant of *Penicillium* sp.

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In our previous studies, four stevastelin congeners named A, B, B3 and C3 have been isolated from culture broth of a fungus, *Penicillium*, as immunosuppressants<sup>1)</sup>. Structures of these stevastelins have been elucidated to be cyclic depsipeptides<sup>2,3)</sup>. In the course of ultraviolet treatment of the strain, we found that a high producing mutant yielded three novel stevastelin congeners. In this report we describe isolation, physico-chemical properties, structural elucidation and immunosuppressive activities of these novel stevastelin congeners named A3, D3 and E3.

In order to isolate a high producing mutant, spores of the stevastelin producing strain, Penicillium sp. NK374186<sup>1)</sup>, were irradiated with a UV lamp under a condition giving 1% survival rate. About a hundred survival strains were picked and cultured as described previously<sup>1)</sup>. From the mycelium of each culture, stevastelins were extracted with MeOH, and again extracted with EtOAc after removal of the initial solvent. These extracts were applied to TLC analysis developed with a mixture of CHCl<sub>3</sub>-MeOH, 20:1 (v/v), and stevastelins were detected with toluidine-chlorine color reaction. With this screening, a strain which produced the known stevastelins at levels about five times as much as the original strain was picked. Furthermore, this strain gave additional positive spots at Rf values of 0.06, 0.07 and 0.42, whereas Rf values of the known stevastelins A, B, B3 and C3 are 0.12, 0.35, 0.49 and 0.33, respectively<sup>1)</sup>. Tentatively, the compounds corresponding to these unidentified spots were named stevastelin A3 (Rf=0.07), D3 (Rf=0.06) and E3 (Rf=0.42), respectively.

The high producing mutant was cultured in a 200 liter jar fermenter as described for the original strain<sup>1)</sup>. From mycelia collected from combined 600 liters of the culture broth, the stevastelins were extracted with 300 liters of MeOH. The MeOH extract was diluted with 300 liters of water and applied to a Diaion HP-20 resin (20 liters). After washing with 50% aqueous MeOH, stevastelin A3 and D3 were eluted with 30 liters of 60% aqueous Me<sub>2</sub>CO, and stevastelin E3 was eluted with 30 liters of 80% aqueous Me<sub>2</sub>CO, stepwisely. The mixture of stevastelin A3 and D3 were applied to Diaion CHP20P resin (2 liters) and eluted with a linear gradient from 50% to 70% aqueous Me<sub>2</sub>CO in order to separate stevastelin A3 and D3. These stevastelins were further purified by Sephadex LH-20 column chromatogaraphy (CHCl<sub>3</sub> - MeOH = 1:1), respectively, to give 800 mg of stevastelin A3 and 130 mg of stevastelin D3 as a colorless powder. On the other hand stevastelin E3 was purified by two successive silica gel (CHCl<sub>3</sub> - MeOH = 50:1 and *n*-hexane - Me<sub>2</sub>CO = 3:1) and Sephadex LH-20 (CHCl<sub>3</sub> - MeOH = 1:1) column chromatography to give 500 mg as a colorless powder.

Stevastelin A3, D3 and E3 are soluble in  $CHCl_3$ , EtOAc, and MeOH. Spectroscopic data of FAB-MS and NMR analysis and other physico-chemical properties of stevastelin A3, D3 and E3 are shown in Table 1. These data of stevastelin A3, D3 and E3 are very similar to that of stevastelin B3<sup>2</sup> (Table 1), confirming that these compounds are stevastelin congeners.

Structure of stevastelin B3 was reported to be composed of four moieties, O-acetyl-L-serine, L-threonine, L-valine and 3,5-dihydroxy-2,4-dimethylstearic  $acid^{2,3}$ (Fig. 1, Table 1). NMR data of stevastelin A3 is similar to stevastelin B3 with regard to O-acetylserine, valine and the fatty acid moieties (Table 1), indicating the presence of these moieties. However, the T3 carbon of stevastelin A3 is shifted downfield ( $\delta_{\rm C}$  70.60, Table 1) when compared with that of stevastelin B3 ( $\delta_{\rm C}$  65.13, Table 1), and the signal due to the T3 hydroxy group is not detected (Table 1). These results suggest a substitution of the T3 hydroxy group of stevastelin A3. Molecular formulas of stevastelin A3 and B3 show that stevastelin A3 has one more sulfur atom and three more oxygen atoms than stevastelin B3 (Table 1). Based on these data, the T3 hydroxy group of stevastelin A3 is





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	B3°		A3		D3		E3		
NMR data <sup>a</sup> Carbon No. <sup>b</sup>	C ppm	H ppm mH ( <i>J</i> = (Hz))	C ppm	H ppm mH (J=(Hz))	C ppm	H ppm mH $(J=(Hz))$	C ppm	H ppm mH $(J=(Hz))$	
[Fatty acid]									
FI	170.43 s		170.74 <sup>d</sup> s		171.23° s	_	171.45 <sup>f</sup> s	-	
F2	41.01 d	2.90 m	41.67 d	2.87 m	40.79 d	2.99 m	40.60 d	3.01 m	
F3	80.13 d	4.91 m	79.62 d	4.90 m	81.19 d	4.99 m	81.59 d	5.00 m	
F4	38.78 d	1.73 m	39.10 d	1.62 m	39.20 d	1.90 m	39.20 d	1.96 m	
F5	68.85 d	3.84 br	68.70 d	3.84 br	68.85 d	3.85 br	68.86 d	3.86 br	
		-OH, 4.37 d (5.8)		–OH, 4.32 m		–OH, 4.44 m		-OH, 4.45 m	
F6	34.79 t	1.20 m	34.81 t	1.20 m	34.84 t	1.36 2H br	34.84 t	1.36 2H br	
		1.34 m		1.32 m					
<b>F</b> 7	25.62 t	1.24 2H	25.63 t	1.24 2H	25.60 t	1.24 2H	25.59 t	1.24 2H	
F8	28.6	1.24 16H	28.6	1.24 16H	28.6	1.24 16H	28.6	1.24 16H	
~15	$\sim 29.0 \text{ t} \times 8$		∼29.0 t ×8		~29.0 t ×8		$\sim 29.0 \times 8$		
F16	31.21 t	1.24 2H	31.21 t	1.24 2H	31.21 t	1.24 2H	31.21 t	1.24 2H	
F17	22.01 t	1.24 2H	22.00 t	1.24 2H	22.00 t	1.24 2H	22.00 t	1.24 2H	
F18	13.86 q	0.86 t 3H (6.8)	13.85 q	0.85 t 3H (6.7)	14.00 q	0.84 t 3H (6.9)	14.01 q	0.85 t 3H (7.0)	
F19	13.72 q	1.12 d 3H (7.3)	13.57 q	1.11 d 3H (7.0)	13.87 q	1.17 d 3H (7.0)	13.85 q	1.17 d 3H (6.2)	
F20	9.11 q	0.54 d 3H (6.6)	9.03 q	0.52 d 3H (7.0)	9.15 q	0.55 d 3H (6.6)	9.15 q	0.56 d 3H (6.6)	
[Valine]									
Vì	170.76 s		170.02 <sup>d</sup> s	-	170.08° s		169.98 <sup>r</sup> s		
V2	61,51 d	4.03 m	61.13 d	3.98 m	61.23 d	4.14 m	61.33 d	4.15 m	
		–NH, 7.67 br		–NH, 7.28 d (10.3)		–NH, 7.97 d (9.9)		–NH, 7.90 d (9.2)	
V3	28.63 d	2.05 m	27.88 đ	2.03 m	28.60 d	2.04 m	29.11 d	2.05 m	
V4a	18.87 q	0.90 d 3H (6.6)	18.32 q	0.85 d 3H (6.6)	18.13 q	0.88 d 3H (6.6)	19.12 q	0.89 d 3H (6.2)	
V4b	19.11 q	0.94 d 3H (6.6)	18.83 q	0.90 d 3H (6.6)	19.03 q	0.93 d 3H (6.6)	20.51 q	0.94 d 3H (6.6)	
[Threonine]									
<b>T</b> 1	170.58 s		169.42 <sup>d</sup> s		167.35° s		168.84 <sup>t</sup> s		
T2	59.30 d	4.16 m -NH, 7.72 d (9.9)	60.51 d	3.94 m -NH, 8.23 d (9.5)	57.29 d	4.35 m NH, 8:15 d (9.2)	57.48 d	4.31 d (9.5) NH, 8.25 d (9.5)	
<b>T</b> 3	65.13 d	4.03 m OH 5 24 d (4 4)	70.60 d	4.32 m	70.58 d	4.88 m	65.26 d	4.50 m OH, 5.28 d (4.0)	
Τ4	20.84 a	1.08 d 3H (6.2)	19.25 g	1.17 d 3H (6.2)	19.16 a	1.21 d 3H (6.2)	19.12 a	1.04 d 3H (6.2)	
[O-Acetvlseri	nel						1		
S1	168.72 s		168.85 <sup>d</sup> s		162.09 <sup>e</sup> s		161.67 <sup>f</sup> s		
S2	50.06 d	4.91 m	49.35 d	4.92 m	131.53 s	_	131.58 s	_	
		-NH, 7.49 d (8.8)		–NH, 7.94 d (7.0)		–NH, 8.30 s		–NH, 8.29 s	
S3	63.15 t	4.20 m	63.21 t	4.25 m	109.76 t	5.77 s	107.96 t	5.74 s	
		4.35 m		4.33 m		6.27 s		6.39 s	
S4	170.04 s		169.82 <sup>d</sup> s	_	_	_			
<b>S</b> 5	20.55 q	2.01 s 3H	20.45 q	2.00 s 3H			·		
EAB MS m/									
Frad-wis $m/z$ Found 5 678 (M + Na) <sup>+</sup>			734 (M -	724 (M H)=		$674 (M - H)^{-1}$		594 $(M - H)^{-}$	
round	round oro (miring)		734 (M - R) 780 (M $\pm 2N_2 - H)^+$		$770 (M + 2N_2 - H)^+$		$596 (M + H)^+ 618 (M + N_2)^+$		
mol			700 (m 1 )	21(4 11)	720 (141 ]	2114 11)	550 (IM	(i), 010 (iii / 14a)	
formula	Ca.H. N.	0.	C. H. N	0.5	Ca.HN	-O.S	C <sub>10</sub> H <sub>44</sub> N	-0-	
mol weight	655.85	- y	735 91	735 91		675.86		595 79	
UV 2MeOH	500100				0.0.00				
(E1%) —					242 nm (96.64)		242 nm (105.21)		
(=					- (-				

### Table 1. Physico-chemical properties of stevastelins.

<sup>a</sup> Recorded in DMSO- $d_6$ .

<sup>b</sup> Refered to Fig. 1.

<sup>c</sup> Cited from ref. 2.

<sup>d~f</sup> These assignments are exchangable.

concluded to be O-sulfonylated (Fig. 1).

NMR data of stevastelin D3 are similar to that of stevastelin B3 with regard to valine and the fatty acid moieties (Table 1). The data of the threonine moiety of stevastelin D3 is also similar to that of stevastelin A3 (Table 1). These data indicate the presence of valine, the fatty acid and *O*-sulfonylthreonine moieties in stevastelin D3 (Fig. 1). However, in the <sup>1</sup>H NMR spectrum of stevastelin D3, signals due to the acetyl group of the *O*-acetylserine moiety are not detected (S4,5, Table 1) and signals due to olefinic protons ( $\delta_{\rm H}$  5.77 [s], 6.27 [s],

Table 1) are newly detected. A presence of olefinic protons is supported by <sup>13</sup>C NMR spectrum detected at  $\delta_{\rm C}$  109.76 [t] and  $\delta_{\rm C}$  131.53 [s] (S3 and S2, Table 1). Also a singlet amide proton ( $\delta_{\rm H}$  8.29 [s], Table 1) and carbonyl carbon ( $\delta_{\rm C}$  162.09 [s], S1, Table 1) are detected in the spectrum of stevastelin D3. From these data stevastelin D3 is concluded to have a dehydro- $CH_2$ 

alanine moiety, -NH-C-CO-, instead of the O-acetylserine moiety (Fig. 1).

NMR data of stevastelin E3 are similar to stevastelin

Table 2. Effect of stevastelin congeners on human T lymphocyte blastogenesis<sup>a</sup>.

Compound	IC <sub>50</sub> (µg/ml)			
B3	0.42 <sup>b</sup>			
A3	30.0			
D3	28.0			
E3	1.8			

<sup>a</sup> Human T lymphocytes were isolated from peripheral blood by density gradient centrifugation. Cells were collected and washed twice with PBS and inoculated into 96-well microplate with 10% FCS-RPMI1640 medium containing 10 mM HEPES buffer. I hour after sample addition, OKT-3 antibody was added at a final concentration of 2 ng/ml. T cell growth was monitored at 3 days with <sup>3</sup>H-thymidine uptake activity.

<sup>b</sup> Cited from ref. 1.

B3 regarding the fatty acid, valine and threonine moieties (Table 1), and are also similar to stevastelin D3 regarding the dehydroalanine moiety (Table 1). These results indicate the structure of stevastelin E3 with these moieties as presented (Fig. 1).

Cyclic structures of the stevastelins were reporterd to be formed by ester linkages between a carboxylic group of the *O*-acetylserine and the F5-hydroxy group in stevastelin B<sup>2)</sup>, and the F3-hydroxy group in stevastelin B3 (Fig. 1)<sup>2)</sup>. Chemical shifts of the F3 carbons of stevastelin A3, D3 and E3 ( $\delta_C$  79.62, 81.19, 81.59, respectively, Table 1) are lower than those of the F5 carbons ( $\delta_C$  68.70, 68.85, 68.86, respectively, Table 1). Furthermore, the F3 hydroxy groups of the fatty acid moieties were not detected in the NMR spectra of stevastelin A3, D3 and E3, whereas the F5 hydroxy groups were detected (Table 1). These results are quite similar to stevastelin B3 (Table 1), and show that ester linkages are formed with the F3 hydroxy groups (Fig.1) as described previously<sup>2)</sup>.

Immunosuppressive activities of stevastelin A3, D3 and E3 against OKT-3 stimulated human T cell proliferation are shown in Table 2. Stevastelin A3 and D3 show weak activities, whereas stevastelin E3 shows potent activity. This result suggests that the O-sulfonylation of the T3 hydroxy group drastically weaken immunosuppressive activity against T cell blastogenesis. On the other hand, an alteration of the O-acetylserine moiety to the dehydroalanine moiety seems to be less influential on the immunosuppressive activity of stevastelin (Table 2).

In this report novel three stevastelin congeners, A3, D3 and E3, were isolated from a cultured broth of the high producing mutant, based on the color reaction to toluidine-chlorine reagent. A congener, stevastelin E3, showed comparable immunosuppressive activity to the known stevastelin congener<sup>1)</sup>. Stevastelin A3 and D3 showed weak activities; however, these compounds were useful to investigate the structure-activity relationship of stevastelin. Another stevastelin congeners with modified amino acid moieties are constructed by chemical conversions (unpublished result). Precise structure-activity relationship of stevastelin will be examined with these congeners and will be reported elsewhere.

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